

A short synthesis of lennoxamine via ynamides

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Abstract—Lennoxamine was synthesized in eight steps from 2,3-dimethoxybenzoic acid via an intermediate ynamide by using palladium-catalyzed Heck–Suzuki–Miyaura domino reactions.

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Polycyclic nitrogen containing heterocycles are encountered in naturally occurring alkaloids and numerous physiologically active drugs.¹ Lennoxamine, an isoindolobenzazepine alkaloid, belongs to the *aporhoedane* series and was extracted from the Chilean plant *Berberis darwinii* (Fig. 1).² Although this compound has no important biological activity, its unique structural feature, five- and seven-membered rings fused with an aromatic moiety, has rendered this molecule attractive as a synthetically challenging target. Several total syntheses of lennoxamine have been reported relying on the construction of ring B or/and ring C as the key steps.³

Recently, we have shown that 3-(arylmethylene)isoindolin-1-ones of type **B** could be obtained from ynamides of type **A** by using palladium-catalyzed Heck–Suzuki–Miyaura domino reactions in the presence of an arylboronic acid (Scheme 1).⁴

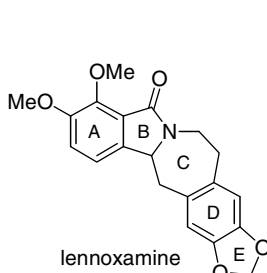
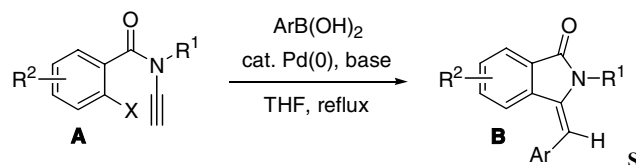


Figure 1. Structure of lennoxamine.

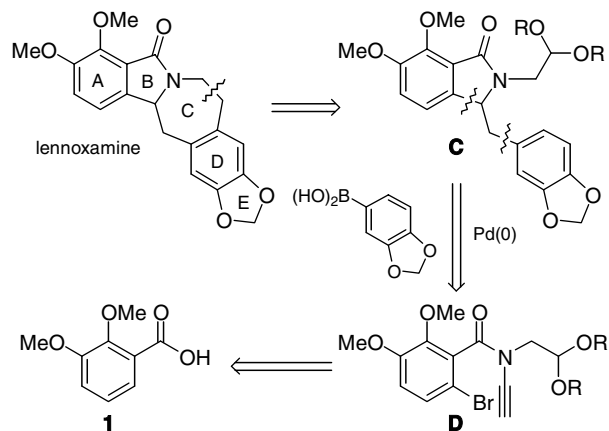
Keywords: Lennoxamine; Isoindolinones; Isoindolobenzazepines; Suzuki–Miyaura coupling; Palladium-catalyzed reactions.

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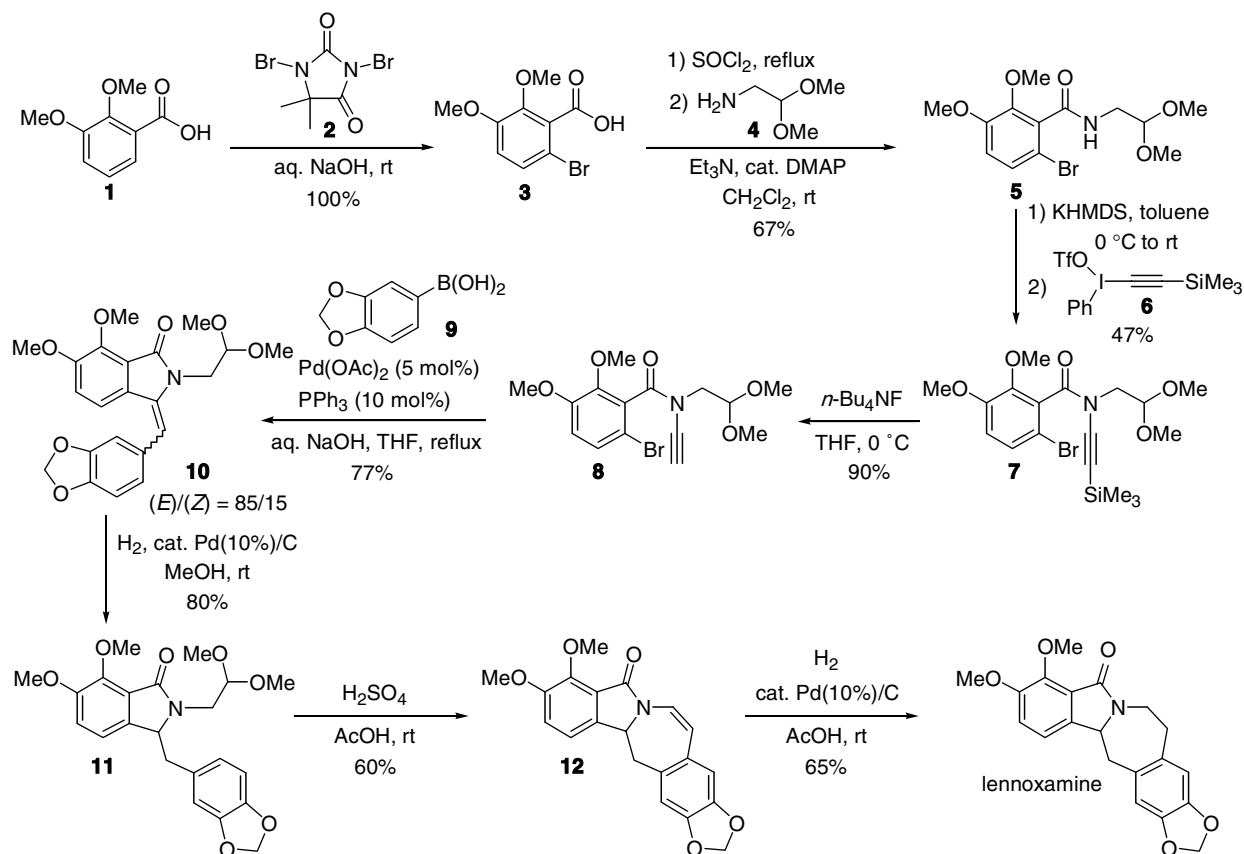


Scheme 1. Synthesis of 3-(arylmethylene)isoindolin-1-ones by Heck–Suzuki–Miyaura domino reactions.

As lennoxamine can arise from the cyclization of an isoindolinone of type **C** under acidic conditions followed by catalytic hydrogenation,^{3i,p} it was envisaged to apply the Heck–Suzuki–Miyaura reactions to an ynamide of type **D**. The latter compound should be prepared from 2,3-dimethoxybenzoic acid **1** (Scheme 2).



Scheme 2. Retrosynthetic analysis of lennoxamine.



Scheme 3. Synthesis of lennoxamine.

The synthesis of lennoxamine (Scheme 3) began with the bromination of 2,3-dimethoxybenzoic acid **1** with 1,3-dibromo-5,5-dimethylhydantoin **2** (aqueous NaOH, rt),⁵ which led to 2,3-dimethoxy-6-bromobenzoic acid **3** in quantitative yield. Carboxylic acid **3** was converted to the corresponding acyl chloride (SOCl_2 , reflux) and the latter was coupled with aminoacetaldehyde dimethyl acetal **4** (Et_3N , cat. DMAP, CH_2Cl_2 , rt) to afford the secondary amide **5** in 67% overall yield (one-pot process). After the formation of the potassium amide (KHMDS, toluene, 0°C to rt) and condensation with the alkyneiodonium salt **6**, ynamide **7** was obtained in modest yield (47%). Alternative alkylation procedures did not provide satisfactory results in the case of amide **5**.⁷ The trimethylsilyl group was then removed by treatment of **7** with tetrabutylammonium fluoride in THF to afford the terminal ynamide **8** (90%). The next task, that constitutes the key step of our synthetic plan, was to construct ring B and hence elaborate the isoindolone core of lennoxamine by the palladium-catalyzed Heck–Suzuki–Miyaura domino reactions.⁴ Thus, the terminal ynamide **8** was treated with a catalytic amount of $\text{Pd}(\text{OAc})_2$ (5 mol%), PPh_3 (10 mol%) and a base (aqueous NaOH) in the presence of the commercially available boronic acid **9** that incorporates both the D and E rings of the natural product (THF, reflux). Under these conditions, the Heck–Suzuki–Miyaura domino reactions proceeded smoothly and the resulting 3-(arylmethylene)isoindolin-1-one **10** was isolated in 77% yield but as a mixture of two geometric isomers

[(E):(Z) = 85:15].^{8,9} The latter result was in sharp contrast with our initial observation that 3-(arylmethylene)isoindolin-1-ones of type **B** were obtained with high (E) stereoselectivity by Heck–Suzuki–Miyaura domino reactions from ynamides of type **A**.⁴ Carbopalladations of alkynes are known to involve a *syn*-addition process but isomerization of the initially formed σ -vinylpalladium complex has been observed in some cases, presumably through a zwitterionic palladium–carbene complex.¹⁰ Although such a scenario cannot be ruled out in the carbopalladation of the triple bond of ynamide **8**, the particular structure of 3-(arylmethylene)isoindolin-1-one **10**, which possesses a rather electron-rich enamide moiety, may be responsible for the observed isomerization. The latter could occur through a reversible protonation of the double bond or a hydration–dehydration pathway.^{11–13} A photochemical process could also be involved due to the light sensitivity of compound **10**.

However, this lack of stereoselectivity in the synthesis of compound **10** had no consequence for the total synthesis of lennoxamine since the mixture of geometric isomers was subsequently subjected to a catalytic hydrogenation (cat. Pd (10%)/C, 1 atm H_2 , MeOH, rt) that led to 3-(arylmethyl)isoindolin-1-one **11** in 60% yield. Completion of the total synthesis could be achieved from compound **11**, as previously reported,^{3i,3p} by treatment under acidic conditions (H_2SO_4 in AcOH, rt) that served to elaborate the seven-membered ring (ring C)³¹ and the

resulting dehydrolennoxamine **12** (60%) was finally hydrogenated (cat. Pd (10%)/C, 1 atm H₂, AcOH, rt) to afford lennoxamine in 65% yield (Scheme 3). The spectroscopic and analytical data of this compound were in perfect agreement with those reported in the literature.³

A concise route to lennoxamine has been developed from 2,3-dimethoxybenzoic acid (8 steps, 7% overall yield) using a strategy complementary to the previously reported approaches for the construction of the isoindolone core of this isoindolobenzazepine alkaloid, that relies on palladium-catalyzed Heck–Suzuki–Miyaura domino reactions from an intermediate ynamide.

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- The ratio of the two geometric isomers was determined by ¹H NMR and their configuration was readily assigned by comparison with the data reported for the (*E*)-isomer.³¹ It was checked that purification of compound **10** by chromatography on silica gel did not lead to any isomerization since the crude product was already an 85/15 isomeric mixture.
- As alkynylsilanes are known to be deprotected under alkaline conditions, the preparation of **10** could also be directly achieved from trimethylsilylynamide **7** by initial treatment with the base (aq NaOH, THF, reflux) followed by addition of the palladium catalyst and arylboronic acid **9** although the overall yield of compound **10** was not improved by this one-pot procedure.
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